

An Unusual Reaction of Methylmagnesium Iodide with Cyclohexadienones

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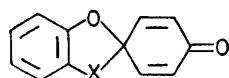
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Summary Reaction of methylmagnesium iodide with the spirocyclohexadienones, (I), gives products consistent with 1,3-addition of the Grignard reagent to the enone system.

It has recently been reported¹ that a minor product of the reaction of *t*-butylmagnesium chloride with ethyl cinnamate is ethyl 2-*t*-butylhydrocinnamate; a radical mechanism is proposed for this reaction. This appears to be the only example cited in the literature of a 1,3-addition of a Grignard reagent to an α,β -unsaturated carbonyl system. We now report that treatment of 1,3-benzodioxole-2-spirocyclohexadien-4'-one² (Ia) with an excess of methylmagnesium iodide, followed by acid hydrolysis of the reaction mixture gives the diphenyl ethers (IIa) and (IIb) as the only products. The identity of these compounds was established by chromatographic and spectroscopic comparison with authentic samples prepared by unambiguous Ullmann ether syntheses. Similarly the reaction of the spiro-sulphonamide (Ib) (obtained by oxidation of *N*-2-hydroxyphenyl-*N*-4-hydroxyphenyl-4-toluenesulphon-

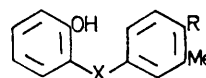
amide) with an excess of methylmagnesium iodide gives in high yield (IIc), again available by an alternative route³ for comparison.

The high specificity of attack by the Grignard at such an unusual reaction site may indicate that the reaction pro-



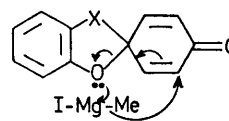
(I)

- (a) X = O
(b) X = NSO₂C₆H₄Me-*p*



(II)

- (a) X = O, R = OH
(b) X = O, R = Me
(c) X = NSO₂C₆H₄Me-*p*
R = Me



(III)

ceeds by a cyclic intermediate⁴ of type (III) to give (IIa). Compounds (IIb) and (IIc) could arise by further 1,2-addition of the Grignard to (III), followed by dehydration of the resulting carbinol. Further studies are in progress to verify this mechanism, and to determine the scope of the

reaction, especially the conditions under which a Grignard might add to an α,β -unsaturated acetal or ketal.

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¹ I. Crossland, *Acta Chem. Scand. B*, 1975, **29**, 468.

² I. G. C. Coutts, D. J. Humphreys, and K. Schofield, *J. Chem. Soc. (C)*, 1969, 1982.

³ I. G. C. Coutts and M. Hamblin, *J.C.S. Perkin I*, in the press.

⁴ M. Cherest, H. Felkin, C. Frajeran, C. Lion, G. Roussi, and G. Swierczewski, *Tetrahedron Letters*, 1966, 875; J. U. Crandall and A. C. Clark, *ibid.*, 1969, 325; H. Felkin, G. Swierczewski, and A. Tambuté, *ibid.*, 1969, 7070. We thank Professor S. J. Huang of the University of Connecticut for useful discussions.